metal-organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 294 K Mean σ (C–C) = 0.005 Å Disorder in main residue R factor = 0.042 wR factor = 0.107 Data-to-parameter ratio = 13.8

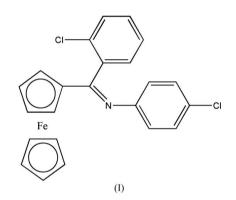
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1-[(2-Chlorophenyl)(4-chlorophenylimino)methyl]ferrocene

The title compound, $[Fe(C_5H_5)(C_{18}H_{12}Cl_2N)]$, a new Schiff base containing a ferrocenyl (Fc) group, has been synthesized and characterized structurally. The compound is a *trans* isomer. The C_{Fc} -C-N-C linkage has a near-planar geometry, indicating conjugation. There are no obvious intermolecular interactions. Received 31 October 2005 Accepted 22 November 2005 Online 7 December 2005

Comment

Some Schiff bases bearing a ferrocenyl group, and their complexes, are excellent non-linear optical materials and liquid crystals (Colbert *et al.*, 1995) because of their strong electron donors and electron-flow bridges. In the course of our investigation of the coordination of Schiff bases with transition metal salts, we observed that the title compound, (I), coordinates readily with Ni^{II} and Cu^{II} salts. The crystal structure of (I) is reported here (Fig. 1).



In (I), the N1–C1 bond length [1.259 (4) Å] confirms that it is a C=N double bond. The torsion angle C6–C10–C11– N1 is 175.1 (3)°, indicating that the imino group is nearly in the plane of the substituted cyclopentadienyl ring. There are no obvious intermolecular interactions.

Experimental

[(4-Anilinylimino)2-chlorophenylmethyl]ferrocene (4.14 g, 0.010 mol) was dissolved in anhydrous acetonitrile (100 ml) and added dropwise to a rapidly stirred mixture of tert-butyl nitrite (1.55 g, 0.015 mol) and anhydrous copper(II) chloride (1.36 g, 0.010 mol) in acetonitrile (20 ml). After 10 h, the reaction mixture was poured into 20% aqueous hydrochloric acid (100 ml). The aqueous acetonitrile mixture was extracted twice with 50 ml portions of ether, the combined ether solution was dried over anhydrous magnesium sulfate, and the ether was removed under reduced pressure. The solid residue was crystallized from ethyl acetate and petroleum ether (1:1) (333–363 K). Analysis calculated for $C_{23}H_{17}Cl_2FeN$: C 63.63, H 3.95, N 3.23%; found: C 63.25, H 4.01, N 3.12%.

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Crystal data

 $[Fe(C_{5}H_{5})(C_{18}H_{12}Cl_{2}N)]$ $M_{r} = 434.13$ Monoclinic, $P_{2_{1}}/n$ a = 8.226 (3) Å b = 17.288 (6) Å c = 13.996 (5) Å $\beta = 103.820$ (6)° V = 1932.8 (11) Å³ Z = 4

Data collection

Bruker SMART CCD area-detector
diffractometer
φ and ω scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\min} = 0.720, \ T_{\max} = 0.880$
10709 measured reflections

Refinement

5	
Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0445P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.042$	+ 0.4751P]
$wR(F^2) = 0.107$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.04	$(\Delta/\sigma)_{\rm max} = 0.001$
3918 reflections	$\Delta \rho_{\rm max} = 0.29 \ {\rm e} \ {\rm \AA}^{-3}$
284 parameters	$\Delta \rho_{\rm min} = -0.26 \ {\rm e} \ {\rm \AA}^{-3}$
H-atom parameters constrained	

 $D_x = 1.492 \text{ Mg m}^{-3}$

Cell parameters from 2747

0.22 \times 0.18 \times 0.12 mm

3918 independent reflections 2546 reflections with $I > 2\sigma(I)$

Mo $K\alpha$ radiation

reflections

 $\theta = 2.8-23.1^{\circ}$ $\mu = 1.06 \text{ mm}^{-1}$

T = 294 (2) K

Prism, red

 $\begin{aligned} R_{\rm int} &= 0.037\\ \theta_{\rm max} &= 26.4^\circ \end{aligned}$

 $h = -7 \rightarrow 10$ $k = -21 \rightarrow 21$ $l = -17 \rightarrow 16$

Table 1

Selected geometric parameters (Å, °).

N1-C11	1.259 (4)	C10-C11	1.469 (4)
N1-C12	1.413 (4)	C11-C18	1.510 (4)
C6-C10	1.431 (4)		
C14 N4 C12	100.0 (0)	N4 C14 C10	100.1 (0)
C11-N1-C12	123.3 (3)	N1-C11-C18	123.1 (3)
N1-C11-C10	119.2 (3)	C17-C12-N1	120.5 (3)
C12-N1-C11-C10	-174.0 (3)	C11-N1-C12-C13	-99.2 (4)
C12-N1-C11-C18	16.9 (5)	N1-C11-C18-C19	81.1 (4)
C6-C10-C11-N1	175.1 (3)		

The C18–C23 benzene ring is disordered over two sites. The two components were refined as rigid ideal hexagons with C–C = 1.39 Å.

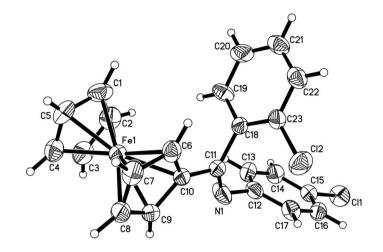


Figure 1

The structure of (I), with the atom-labelling scheme. Displacement ellipsoids are drawn at the 35% probability level. H atoms are shown as small spheres of arbitrary radii. Only one disorder component is shown.

The site occupancy factors are 0.609 (3) and 0.391 (3). All H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C-H = 0.93 or 0.98 Å and $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

References

Bruker (1997). SMART, SAINT and SHELXTL. Bruker AXS Inc., Madison, Wisconsin, USA.

Colbert, M. C. B., Hodgson, D., Lewis, J., Raithby, P. R. & Long, N. J. (1995). Polyhedron, 14, 2759–2766.

Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.

Sheldrick, G. M. (1997). SHELXS97 and SHELXS97. University of Göttingen, Germany.